Cuprous bromide-promoted cyclization of ketene N,S-acetals with prop-2-ynyl bromide: synthesis of regiospecifically substituted and annelated 3-acyl(or nitro)-5-methyl pyrroles

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Acyclic 10a-h and cyclic 14a-b acyl and nitroketene 12a-b N,S-acetals undergo facile cuprous bromide induced cyclization with prop-2-ynyl bromide to afford 1-substituted-3-acyl(or nitro)-5-methyl-2-(methylsulfanyl)pyrroles 11a-h and 13a-b and the corresponding annelated analogues 15a-b in good yields.

Claisen rearrangement of suitable prop-2-ynyl ethers¹ and thioethers^{1,2} are known for both aliphatic and aromatic compounds. The final products isolated in these rearrangements are generally pyran (or benzopyran) and furan (benzofuran) derivatives or the corresponding thio analogues (from thio-Claisen rearrangement). A few reports are available on the related acetylenic aza-Claisen rearrangement which usually require higher temperature or Lewis acid catalysis and afford either pyridine or quinoline derivatives.³ In a recent paper, we described our studies on acetylenic oxa-Claisen rearrangement of oxoketene O-prop-2-ynyl S-methyl acetals to give either diene esters 4 (toluene heating) or the furans 5 (K₂CO₃/Et-COMe) depending on the reaction conditions.⁴ Analogous thio-Claisen rearrangement of the corresponding S-prop-2-ynyl thioacetal 2 (R = Ph) (generated in situ) on the other hand yields the dihydrothiopyran 7 instead of 2-methylthiophene 6 (Scheme 1).⁵ These rearrangements proceed rapidly under relatively mild conditions and thus suggested that the corresponding N-prop-2-ynyl S-methyl acetal 3 might also rearrange to either dihydropyridine or pyrrole derivatives. This prompted us to attempt synthesis of N-prop-2-ynyl S-methyl acetals 3 and study the reaction of N,S-acetals with prop-2-ynyl bromide under various conditions. Although we were not successful in synthesizing the intermediate 3, the corresponding anticipated products of aza-Claisen rearrangement from 3, i.e. 1-substituted-3-acyl-5-methyl-2-(methylsulfanyl)pyrroles 11, could be obtained by cuprous bromide-promoted cyclization of N,S-acetals with prop-2-ynyl bromide. The results of these studies are described in this paper.

Results and discussion

Attempted synthesis of the unsubstituted N-prop-2-ynyl Smethyl acetal 3a by direct displacement on ketene S, S-acetal 8 or the corresponding sulfonium salt 9 by prop-2-ynylamine under different reaction conditions gave either starting material or an intractable mixture of products (Scheme 2). Similarly, our various trial experiments involving base-induced alkylation of N,S-acetal 10a with prop-2-ynyl bromide to give either Nmethyl-N-prop-2-ynyl S-methyl acetal 3b or its subsequent rearrangement products resulted in failure. We therefore treated N,S-acetal 10a with prop-2-ynyl bromide in the presence of various cuprous salts and copper catalysts. Best results were obtained with cuprous bromide and the product (63%) thus isolated was identified as 3-benzoyl-1,5-dimethyl-2-(methylsulfanyl)pyrrole 11a on the basis of its spectral and analytical data. No trace of N-methyldihydropyridine or any other product of acetylenic aza-Claisen rearrangement was observed



Scheme 1 Reagents and conditions: i, toluene, heat; ii, K_2CO_3 , EtCOMe, heat; iii, KOBu^t, DMF



Scheme 2 Reagents and conditions: i, HC=CCH₂NH₂; NaH, DMF; ii, HC=CCH₂Br

in the reaction mixture. The reaction was found to be general and other N-substituted aroyl **10b–f** and acetyl **10g–h** ketene N,S-acetals similarly afforded the corresponding pyrroles **11b–h** in moderate to good yields under identical conditions (Scheme 3). The reaction was equally facile with nitroketene N,S-acetals **12a–b** and the corresponding 5-methyl-3-nitropyrroles **13a–b** could be obtained in 57 and 81% yields, respectively (Scheme 4). The cyclic N,S-acetals **14a–b** also reacted with prop-2-ynyl bromide in a similar fashion under these conditions to afford annelated pyrroles **15a–b** in good yields.



Scheme 3 Reagents and conditions: i, Cu¹Br, dioxane, 100 °C



Scheme 4 Reagents and conditions: i, Cu¹Br, dioxane, 100 °C

The proposed mechanism for the formation of the pyrroles 11, 13 and 15 is outlined in Scheme 5. Apparently, the reaction between N,S-acetals and prop-2-ynyl bromide is catalysed in the presence of Cu'Br, since the reaction between N,S-acetals and prop-2-ynyl bromide was not observed in its absence, with few exceptions where only traces of the product pyrroles were formed. Subsequently, the allene intermediate 16 appears to undergo rapid cyclization by intramolecular nucleophilic attack of the amino group on the reactive allenic central carbon to afford the pyrroles.

The reaction provides a facile regiocontrolled synthesis of 1,2-substituted and annelated 3-acyl(or nitro)-5-methylpyrroles from easily accessible N,S-acetals, although the overall transformation does not appear to involve any aza-Claisen rearrangement.⁶ Our efforts to synthesize substrates for aliphatic acetylenic aza-Claisen rearrangement and to probe the mechanism of their Lewis acid catalysis are in progress.

Experimental

Mps were determined on a Thomas Hoover melting point (capillary method) apparatus and are uncorrected. IR spectra were measured using a Perkin-Elmer 297 spectrometer. ¹H NMR (90 MHz) spectra were recorded on a Varian EM-390 spectrometer. High field ¹H NMR (300 MHz) and ¹³C NMR (75 MHz) spectra were recorded on a Gemini 300 BB spectrometer. Chemical shifts are reported in ppm relative to internal tetramethylsilane and J values in Hz. Mass spectra were measured on a JEOL JMS-D 300 mass spectrometer. Elemental analysis was carried out on a Heraeus CHN-O-Rapid Analyser. Dry dioxane was freshly distilled from sodium benzophenone ketyl prior to use.



General procedure for the preparation of 1,2,3-trisubstituted-5methylpyrroles 11a-h, 13a-b and 15a-b

To a solution of prop-2-ynyl bromide (1.18 g, 10 mmol) in dry dioxane (30 ml) was added Cu^IBr (1.43 g, 10 mmol) in one portion. After the mixture had been stirred at ambient temperature for 0.5 h, the corresponding *N*,*S*-acetal (10 mmol) was added and heated under reflux for 4–6 h (monitored by TLC). The cold reaction mixture was poured into ice cold water (50 cm³) and extracted with chloroform (2×50 cm³). The combined extracts were dried (Na₂SO₄) and evaporated to give crude pyrroles which were chromatographed on silica gel eluting with hexane–ethyl acetate (20:1).

3-Benzoyl-1,5-dimethyl-2-(methylsulfanyl)pyrrole 11a. Viscous oil (1.54 g, 63%); $v_{max}(neat)/cm^{-1}$ 1638 (CO); $\delta_{H}(90 \text{ MHz}; \text{CCl}_4)$ 2.17 (3 H, s, CH₃), 2.33 (3 H, s, SCH₃), 3.63 (3 H, s, NCH₃), 6.03 (1 H, s, 4-H), 7.26–7.46 (3 H, m, ArH) and 7.66–7.80 (2 H, m, ArH); $\delta_{C}(75 \text{ MHz}; \text{CDCl}_3)$, 13.03 (CH₃), 20.33 (SCH₃), 30.94 (NCH₃), 111.22, 127.94, 129.52, 131.47 (C-4, C-5', C-2', C-3', C-4' and C-6') and 131.01, 126.44, 128.05, 140.13 and 191.47 (C-2, C-3, C-5, C-1' and CO); m/z 245 (M⁺, 100%), 230 (41) and 212 (65) (Found: C, 68.5; H, 6.15; N, 3.5. C₁₄H₁₅NOS requires C, 68.54; H, 6.16; N, 5.71%).

3-Benzoyl-5-methyl-2-methylsulfanyl-1-propylpyrrole 11b. Viscous oil (1.34 g, 49%); $v_{max}(neat)/cm^{-1}$ 1638 (CO); $\delta_{H}(90 \text{ MHz}; \text{CCl}_{4})$ 0.96 (3 H, t, J7, CH₂CH₃), 1.74 (2 H, distorted sext, J7, CH₂CH₃), 2.23 (3 H, s, CH₃), 2.50 (3 H, s, SCH₃), 4.00 (2 H, t, J7, NCH₂), 6.06 (1 H, s, 4-H), 7.26–7.53 (3 H, m, ArH) and 7.70–7.90 (2 H, m, ArH); *m*/z 273 (M⁺, 100%), 258 (33) and 240 (72) (Found: C, 69.2; H, 6.5; N, 5.25. C₁₆H₁₉NOS requires C, 70.29; H, 7.00; N, 5.12%).

3-(4-Chlorobenzoyl)-1-isopropyl-5-methyl-2-(methylsulfanyl)pyrrole 11c. Viscous oil (2.43 g, 79%); $v_{max}(neat)/cm^{-1}$ 1647 (CO); $\delta_{H}(90 \text{ MHz}; \text{CCl}_{4})$ 1.53 [6 H, d, J 7, CH(CH₃)₂], 2.28 (3 H, s, CH₃), 2.35 (3 H, s, SCH₃), 5.14 [1 H, distorted sext, J 7, CH(CH₃)₂], 6.00 (1 H, s, 4-H), 7.30 (2 H, d, J 9, ArH) and 7.72 (2 H, d, J 9, ArH); m/z 308 (M⁺, 25%), 307 (M⁺, 100), 292 (22) and 274 (38) (Found: C, 62.3; H, 5.7; N, 4.7. C₁₆H₁₈ClNOS requires C, 62.43; H, 5.89; N, 4.55%.

3-Benzoyl-1-butyl-5-methyl-2-(methylsulfanyl)pyrrole 11d. Viscous oil (1.46 g, 51%); $\nu_{max}(neat)/cm^{-1}$ 1642 (CO); $\delta_{H}(90$ MHz; CCl₄) 0.96 (3 H, t, J 7, CH₃), 1.20–1.66 (4 H, distorted quint, J7, 2 × CH₂CH₃), 2.20 (3 H, s, CH₃), 2.40 (3 H, s, SCH₃), 4.09 (2 H, t, J 7, NCH₂), 6.13 (1 H, s, 4-H), 7.41–7.56 (3 H, m, ArH) and 7.76–7.91 (2 H, m, ArH); m/z 287 (M⁺, 47%), 272 (13) and 254 (34) (Found: C, 71.0; H, 7.35; N, 4.9. C₁₇H₂₁NOS requires C, 71.04; H, 7.36; N, 4.67%).

1-Benzyl-5-methyl-3-(4-methylbenzoyl)-2-(methylsulfanyl)pyrrole 11e. Viscous oil (2.15 g, 64%); $v_{max}(neat)/cm^{-1}$ 1688 (CO); $\delta_{H}(90 \text{ MHz}; \text{CCl}_4)$; 2.10 (3 H, s, CH₃), 2.25 (3 H, s, CH₃), 2.40 (3 H, s, SCH₃), 5.34 (2 H, s, NCH₂), 6.14 (1 H, s, 4-H), 6.90 (2 H, d, J 9, ArH), 7.10–7.30 (5 H, m, ArH) and 7.73 (2 H, d, J 9, ArH); m/z 335 (M⁺, 52%) and 302 (19) (Found: C, 75.1; H, 6.2; N, 4.4. C₂₁H₂₁NOS requires C, 75.19; H, 6.31; N, 4.18%).

3-Benzoyl-5-methyl-2-methylsulfanyl-1-phenylpyrrole 11f. Colourless crystals (1.66 g, 54%); mp 118 °C (from chloroformhexane); v_{max} (KBr)/cm⁻¹ 1638 (CO); δ_{H} (90 MHz; CCl₄), 2.05 (3 H, s, CH₃), 2.20 (3 H, s, SCH₃), 6.20 (1 H, s, 4-H), 7.13–7.33 (3 H, m, ArH), 7.34–7.56 (5 H, m, ArH) and 7.73–7.93 (2 H, m, ArH); *m*/*z* 307 (M⁺, 100%), 292 (31) and 274 (78) (Found: C, 74.0; H, 5.3; N, 4.8. C₁₉H₁₇NOS requires C, 74.23; H, 5.58; N, 4.56%).

3-Acetyl-1-ethyl-5-methyl-2-(methylsulfanyl)pyrrole 11g. Viscous oil (1.02 g, 52%); $v_{max}(neat)/cm^{-1}$ 1660 (CO); $\delta_{H}(90 \text{ MHz}; \text{CCl}_4)$, 1.26 (3 H, t, J 7, CH_2CH_3), 2.21 (3 H, s, CH_3), 2.30 (3 H, s, CH_3CO), 2.34 (3 H, s, SCH_3), 4.06 (2 H, q, J7, CH_2CH_3) and 6.20 (1 H, s, 4-H); m/z 197 (M⁺, 100%), 182 (61) and 164 (63) (Found: C, 60.7; H, 7.5; N, 7.3. $C_{10}\text{H}_{15}\text{NOS}$ requires C, 60.87; H, 7.66; N, 7.10%).

3-Acetyl-1-benzyl-5-methyl-2-(methylsulfanyl)pyrrole 11h. Viscous oil (1.40 g, 54%); $v_{max}(neat)/cm^{-1}$ 1663 (CO); $\delta_{H}(90 \text{ MHz}; \text{CCl}_4)$, 2.12 (3 H, s, CH₃), 2.16 (3 H, s, COCH₃), 2.44 (3 H, s, SCH₃), 5.33 (2 H, br s, CH₂), 6.36 (1 H, s, 4-H), 6.77–6.95 (2 H, m, ArH) and 7.16–7.34 (3 H, m, ArH); *m/z* 259 (M⁺, 59%) and 226 (11) (Found: C, 69.2; H, 6.5; N, 5.25. C₁₅H₁₇NOS requires C, 69.46; H, 6.61; N, 5.40%).

1-Ethyl-5-methyl-2-methylsulfanyl-3-nitropyrrole 13a. Viscous oil (1.14 g, 57%); $v_{max}(neat)/cm^{-1}$ 1500 and 1450; $\delta_{H}(90 \text{ MHz}; \text{CCl}_4)$, 1.29 (3 H, t, J7, CH₂CH₃), 2.24 (3 H, s, CH₃), 2.39 (3 H, s, SCH₃), 4.13 (2 H, q, J7, CH₂CH₃) and 6.43 (1 H, s, 4-H); m/z 200 (M⁺, 100%) and 153 (34) (Found: C, 47.8; H, 6.15; N, 14.0. C₈H₁₂N₂O₂S requires C, 47.98; H, 6.04; N, 13.99%).

1-Benzyl-5-methyl-2-methylsulfanyl-3-nitropyrrole 13b. Viscous oil (2.12 g, 81%); v_{max} (neat)/cm⁻¹ 1505 and 1450; δ_{H} (90 MHz; CCl₄), 2.13 (3 H, s, CH₃), 2.26 (3 H, s, SCH₃), 5.35 (2 H, s, NCH₂), 6.55 (1 H, s, 4-H), 6.80–7.00 (2 H, m, ArH) and 7.14–7.34 (3 H, m, ArH); *m*/*z* 262 (M⁺, 52%) (Found: C, 59.3; H, 5.3; N, 10.8. C₁₃H₁₄N₂O₂S requires C, 59.52; H, 5.38; N, 10.68%).

7-(4-Chlorobenzoyl)-5-methyl-2,3-dihydropyrrolo[2,1-b]thiazole 15a. Viscous oil (1.80 g, 65%); $v_{max}(neat)/cm^{-1}$ 1615 (CO); $\delta_{\rm H}(90 \text{ MHz}; \text{CCl}_4)$, 2.33 (3 H, s, CH₃), 3.23 (2 H, t, J 9, SCH₂), 4.33 (2 H, t, J9, NCH₂), 6.26 (1 H, s, 4-H), 7.30 (2 H, d, J 9, ArH) and 7.96 (2 H, d, J9, ArH); m/z 278 (M⁺, 39%) and 276 (100) (Found: C, 60.35; H, 4.65; N, 5.3. C₁₄H₁₂NOSCI requires C, 60.53; H, 4.36; N, 5.04%).

 7-Acetyl-5-methyl-2,3-dihydropyrrolo[2,1-b]thiazole
 15b.

 Viscous oil (1.03 g, 57%); $v_{max}(neat)/cm^{-1}$ 1615 (CO); $\delta_{H}(90$ MHz; CCl₄), 2.20 (3 H, s, CH₃), 2.47 (3 H, s, CH₃CO), 3.22 (2 H, t, J 9, SCH₂), 4.24 (2 H, t, J 9, NCH₂) and 6.06 (1 H, s, 4-H); m/z 181 (M⁺, 100%) and 121 (97.3) (Found: C, 59.4; H, 6.0; N, 8.0. C₉H₁₁NOS requires C, 59.64; H, 6.12; N, 7.73%).

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References

- (a) Review: A. Viola, J. J. Collins and N. Filipp, *Tetrahedron*, 1981, 37, 3765; (b) A. Jefferson and F. Scheinmann, *Quart. Rev. Chem.* Soc., 1968, 22, 391; (c) R. Gericke and I. Lues, *Tetrahedron Lett.*, 1992, 33, 1871; (d) B. A. Otter, S. S. Saluja and J. J. Fox, *J. Org. Chem.*, 1972, 37, 2858; (e) J. K. Crandall and G. L. Tindell, *J. Chem.* Soc., *Chem. Commun.*, 1970, 1411; (f) M. R. Attwood, I. Churcher, R. M. Dunsdon, D. N. Hurst and P. S. Jones, *Tetrahedron Lett.*, 1991, 32, 811.
- 2 (a) H. Kwart and J. J. George, J. Chem. Soc., Chem. Commun., 1970, 433; (b) L. Dalgaard and S.-O. Lawesson, Tetrahedron, 1972, 28, 2051; (c) B. W. Bycraft and W. Landon, J. Chem. Soc., Chem. Commun., 1970, 168; (d) J. Meyer and L. Brandsma, Recl. Trav. Chim. Pays-Bas, 1972, 91, 578.
- 3 (a) K. Berg-Nielson and L. Skattebol, Acta Chem. Scand., Ser. B, 1978, **32**, 553; (b) R. D. Dillard, D. E. Pavey and D. N. Benslay, J. Med. Chem., 1973, **16**, 251; (c) N. R. Easton and D. R. Cassades, J. Org. Chem., 1962, **27**, 4713; (d) G. R. Cook, N. S. Barta and J. R. Stille, J. Org. Chem., 1992, **57**, 461.
- 4 L. N. Bhat, H. Ila and H. Junjappa, J. Chem. Soc., Perkin Trans. 1, 1994, 1749.
- 5 F. C. V. Larsson and S.-O. Lawesson, Tetrahedron, 1972, 28, 5341.
- 6 (a) A. Kumar, H. Ila and H. Junjappa, J. Chem. Soc., Chem. Commun., 1976, 593; (b) A. K. Gupta, R. T. Chakrasali, H. Ila and H. Junjappa, Synthesis, 1989, 141; (c) A. J. G. Baxter, J. Fuher and S. J. Teague, Synthesis, 1994, 207; (d) T. D. Lash, J. R. Bellettini, J. A. Bastian and K. B. Couch, Synthesis, 1994, 170.

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